

Tetrahedron 56 (2000) 2225-2230

Stereoselective Route towards 2,5-Disubstituted Piperidine Alkaloids. Synthesis of $(+)$ -Pseudoconhydrine and (\pm) -epi-Pseudoconhydrine

Joakim Löfstedt,^a Helena Pettersson-Fasth^b and Jan-E. Bäckvall^{a,*}

a Department of Organic Chemistry, Arrhenius Laboratory, Stockholm University, SE-106 91 Stockholm, Sweden ^bDepartment of Organic Chemistry, Uppsala University, Box 531, SE-751 21 Uppsala, Sweden

Accepted 7 December 1999

Abstract—This paper describes a new general approach towards functionalized piperidine alkaloids, based on the stereo- and regioselective palladium(0)-catalyzed nucleophilic ring-opening of vinyl epoxides by nitrogen nucleophiles. The latter reaction provides access to stereodefined anti and syn aminoalcohol derivatives, 1-benzyloxy-5-(p-toluenesulfonamido)-3-alken-2-ols (5), which were transformed to $(+)$ pseudoconhydrine (3) and (\pm) -epi-pseudoconhydrine (9), respectively, via protection (silyl ether), hydrogenation, debenzylation and cyclization. Detosylation-deprotection gave the final products in good yields and high stereoisomeric purity. © 2000 Elsevier Science Ltd. All rights reserved.

Introduction

Piperidine alkaloids are widespread in nature and many compounds of this family exhibit interesting biological activity.¹ Cassein $(1)^2$ isolated from *Cassia excelsa* and spectalin $(2)^3$ from *Cassia spectabilis* are members of this family that have been subject to synthetic studies. Pseudoconhydrine (3) has been isolated along with coniine (4) from Poison Hemlock, Conium maculatum.⁴ Syntheses of compound 3 in its racemic⁵ and enantiomerically pure⁶ form using different strategies have been reported previously.

While there are several known routes towards piperidine alkaloids, most of them suffer from lack of generality. It was therefore of interest to develop new general approaches

that give access to all stereoisomers via a single common route. The strategy outlined in the present work provides a route to 2,5-disubstituted piperidine alkaloids of either cis or trans stereochemistry using 1-benzyloxy-5-(p-toluenesulfonamido)-3-alken-2-ols (5) as key intermediates. The absolute stereochemistry is introduced early in the sequence via Sharpless epoxidation.

The use of amidoalkenols with the general structure 5 as building blocks in the synthesis of 2,5-disubstituted pyrrolidine derivatives (6) and indolizidine alkaloids (7) has previously been demonstrated in our laboratory⁷ (Scheme) 1). Amidoalkenols of *anti* (5a) and $syn(5b)$ stereochemistry are readily available by a common route, starting from either cis or trans-1-benzyloxy-2,3-epoxybutan-4-ol (8a or 8b), respectively. The versatility of intermediates 5 is further demonstrated in this paper by the synthesis of $(+)$ -pseudoconhydrine (3) from 5a, and (\pm) -epi-pseudoconhydrine (9) from 5b (Scheme 1).

Results and Discussion

Amidoalkenol 5a with alkyl side chains of varying length is readily available from *trans* epoxyalcohol $\hat{8a}$.^{7a} The key step in the synthesis of $5a$ is the Pd(0)-catalyzed regioand stereoselective ring-opening of vinylepoxide 10a by NaNHTs.⁸ The reaction involves an external attack by $TsNH$ on an intermediate $(\pi$ -allyl)palladium complex.^{7a,8a,9} Protection of the hydroxy group of $5a$ as a TBDMS ether using standard conditions¹⁰ gave the ether 11a in 91% yield, and subsequent hydrogenation employing

Keywords: $(+)$ -pseudoconhydrine; (\pm) -epi-pseudoconhydrine.

^{*} Corresponding author. Tel.: +46-8-6747178; fax: +46-8-154908; e-mail: jeb@organ.su.se

^{0040-4020/00/\$ -} see front matter © 2000 Elsevier Science Ltd. All rights reserved. PII: S0040-4020(99)01105-9

Scheme 1.

Scheme 2. (a) TBDMSCl, imidazole, (b) PtO₂, EtOH, H₂ (c) Pd/C, MeOH, H₂, (d) MsCl, Et₃N, THF, (e) K₂CO₃, MeOH, (f) Na(Hg), Na₂HPO₄, MeOH, (g) HCl, EtOH.

Scheme 3. (a) TBDMSCl, imidazole, (b) PtO₂, EtOH, H₂ (c) Pd/C, MeOH, H₂, (d) MsCl, Et₃N, THF, (e) K₂CO₃, MeOH, (f) Na(Hg), Na₂HPO₄, MeOH, (g) HCl, EtOH.

PtO₂ and 1 atm. of H_2^{11} gave 12a in 97% yield. The benzyl group of $12a$ was removed by hydrogenolysis¹⁰ over palladium on carbon to give 13a in 93% yield. Mesylation and subsequent cyclization employing K_2CO_3 in MeOH¹² afforded the tosyl piperidine 15a in 81% yield. The N -tosyl protective group was removed with Na/NH₃ in EtO H^{13} to give the free amine 16a in 85% yield. Stirring of 16a in EtOH containing 1% HCl (10 equiv.) afforded $(+)$ -3 as its HCl salt in 98% yield.¹⁴ The free base was released by treatment with $2 M$ NaOH (Scheme 2). $(+)$ -Pseudoconhydrine (3) was characterized and spectral data were in accordance with those previously reported.^{6e}

The same reaction sequence was employed to convert the racemic syn aminoalcohol derivative 5b to (\pm) -9 (Scheme 3). The syn amidoalcohol 5b was obtained from palladiumcatalyzed ring opening of the (cis, Z)-epoxide 10b by NaNHTs.^{7a} The detosylation step in this sequence was

Scheme 4. (a) Ag₂O, BnBr, CH₂Cl₂, (b) (-)-DET (7%), Ti(i-PrO)₄ (6%), t-BuOOH, CH₂Cl₂.

carried out using Na(Hg).¹⁵ The (\pm) -epi-pseudoconhydrine (9) was characterized as its hydrochloride salt which is known in the literature.^{6b}

For an efficient synthesis of the target compounds it is necessary to have easy access to 8a and 8b. The latter compounds are prepared from (E) - and (Z) -2-butene-1,4diol, respectively. The (E) -2-butene-1,4-diol (18a) was prepared by reduction of the alkyne moiety in 17 with Li AlH_4 ^{16,17} Monobenzylation of the (Z)-2-butene-1,4-diol under the previously employed standard conditions (1 equiv. NaH, BnBr, THF) worked satisfactorily and afforded the monoether in 70% yield. However, the corresponding monobenzylation of (E) -2-butene-1,4-diol (18a) in THF was sluggish and gave a poor yield of the monoether $(<20\%)$. For an efficient synthesis of pseudoconhydrine $((+)$ -3) it was therefore necessary to improve the synthesis of 8a and hence the monoalkylation of 18a to 18b. A change of solvent to DMF led to some improvement of the yield (50%). Recently, the use of Ag₂O together with BnBr was successfully used to monobenzylate diols.¹⁸ Reaction of 18a with 1.1 equiv. BnBr in the presence of 1.0 equiv. Ag₂O in CH₂Cl₂ (RT)¹⁹ dramatically improved the yield (84%) as well as the simplicity of the reaction.

Sharpless epoxidation²⁰ using standard conditions, 7% (-)diethyltartrate ((-)-DET) and 6% Ti(i -PrO)₄, gave the epoxide as an oil in 87% yield and 93% ee. To improve the enantiomeric excess by recrystallization, the product was converted to the 2,4-dichlorobenzoate. The ester was successfully recrystallized and hydrolyzed (NaOH in MeOH, RT, overnight) to give the epoxide 8a in more than 99.5% ee (Scheme 4).

Conclusions

The piperidine alkaloids $(+)$ -pseudoconhydrine (3) and (\pm) -epi-pseudoconhydrine (9) were synthesized employing 5a and 5b, respectively, as stereodefined key intermediates. By using a slight modification of this method 5-hydroxy-2,6-dialkylpiperidines such as cassein (1) and spectaline (2) should also be available.

The possibility of forming stereodefined 5- and 6-membered ring structures from the same precursor makes amidoalkenols 5 powerful tools for the synthesis of these heterocyclic compounds.

Experimental

General

using chloroform-d₁ (7.26 ppm $\mathrm{^{1}H}$, 77 ppm $\mathrm{^{13}C}$) as internal standard unless otherwise stated. Accurate mass measurements were preformed with a JEOL SX 102 mass spectrometer with m-nitrobenzylalcohol as matrix. IR spectra were obtained on a Perkin-Elmer FTIR spectrometer. Optical rotations were measured using Perkin–Elmer 241 polarimeter. GC-MS were measured with a Thermo Quest GCQ plus. The (Z)-2-butene-1,4-diol and 2-butyne-1,4-diol are commercially available. The enantiomeric purity of epoxide 8a was determined by HPLC analysis using a Chiragel OD-H column, flow rate 0.5 ml/min, 80/20, hexane/2propanol.

Compound 18b. To a stirred solution of the diol 18a $(0.53 \text{ g}, 6.0 \text{ mmol})$ in CH_2Cl_2 (11 ml) was added freshly prepared Ag₂O (1.39 g, 6.0 mmol) and BnBr (0.78 ml, 6.6 mmol). The reaction was stirred for 15 h and filtered through a short silica gel pad, which was washed with diethyl ether. Evaporation of the solvent followed by flash chromatography (SiO₂, pentane: ether (75:25)) gave 0.90 g (84%) of the monoalkylated product 18b. Spectral data were in accordance with those previously reported.¹⁶

Compound 5a was prepared using the procedure previously described.^{7a} Pd(PPh₃)₄ (33 mg, 0.028 mmol), NaNHTs $(110 \text{ mg}, 0.57 \text{ mmol})$, and NH₂Ts $(100 \text{ mg}, 0.57 \text{ mmol})$ were combined and an argon atmosphere was applied. CH₃CN (5 ml) was added and a yellow color appeared. A solution of $10a$ (110 mg, 0.47 mmol) in 0.5 ml CH₃CN was added and the reaction was stirred at 40° C for 6 h. 170 mg (90%) of **5a** was isolated: ¹H NMR (CDCl₃) δ 7.70 (d, $J=8.4$ Hz, 2H, Ts), 7.31 (m, 5H, Ph), 7.21 (d, $J=8.4$ Hz, 2H, Ts), 5.40 (m, 2H, CH=CH), 5.08 (d, $J=8.1$ Hz, 1H, NH), 4.50 (s, 2H, PhCH2), 4.12 (m, 1H, CHOH), 3.72 (quintet, $J=7.3$ Hz 1H, CHNHTs), 3.29 (dd, $J=3.3$, 9.5 Hz, 2H, BnOCH₂), 3.15 (dd, $J=7.7$, 9.5 Hz, 2H, BnOCH₂), 2.52 (d, J=3.3 Hz, 1H, OH), 2.36 (s, 3H, Ts), 1.20 -1.50 (several m, 4H), 0.80 (t, J=7.3 Hz, 3H, Me). ¹³C NMR (CDCl₃) δ 142.7, 138.0, 137.5, 131.4, 129.7, 129.1 (2C), 128.1 (2C), 127.5, 127.4 (2C), 126.9 (2C), 73.7, 73.1, 70.2, 55.2, 37.7, 21.4, 18.5, 13.8. $[\alpha]_D^{23} = -12^{\circ}$ (c=1.15, $CHCl₃$).

Compound 11a. To a stirred solution of amidoalcohol 5a $(0.35 \text{ g}, 0.87 \text{ mmol})$ in DMF (3.5 ml) at 0°C was added imidazole $(0.19 \text{ g}, 2.78 \text{ mmol})$ and TBDMSCl $(0.20 \text{ g},$ 1.30 mmol). The mixture was stirred at 0° C for 2 h and at RT for an additional 2 h. The reaction mixture was then partitioned between ether and brine, then dried (Na_2SO_4) and concentrated. The crude product was purified by flash chromatography (SiO₂, pentane: ether $(1:1)$) to afford 0.41 g, (91%) of the product. ¹H NMR (CDCl₃) δ 7.72 (d, $J=7.9$ Hz, 2H, Ts), 7.34 (m, 5H, Ph), 7.31 (d, $J=7.9$ Hz, 2H, Ts), 5.40 (m, 2H, CH=CH), 4.97 (d, J=7.8 Hz, 1H, NH), 4.48 (d, $J=2.0$, 2H, PhCH₂), 4.15 (m, 1H,

CHOTBDMS), 3.76 (quintet, $J=6.5$ Hz 1H, CHNHTs), 3.19 (m, 2H, BnOCH₂) 2.39 (s, 3H, Ts) 1.20 -1.50 (several m, 4H), 0.85 (m, 12H, t -BuSi+Me), 0.00 (s, 3H, MeSi), -0.03 (s, 3H, MeSi). ¹³C NMR (CDCl₃) $\delta(142.8, 138.2,$ 138.1, 131.7, 130.1, 129.4 (2C), 128.2 (2C), 127.4 (2C), 127.3 (2C), 127.0, 74.9, 73.3, 71.7, 55.1, 38.2, 25.9, 21.6, 18.6, 18.4, 13.8, 24.6. IR 3273, 2929, 2857, 1453, 1161, 1094, 698. HRMS Calcd for $C_{28}H_{43}O_4$ NSiS + Na: 540.2580, found: 540.2597. $[\alpha]_D^{22} = -15^\circ$ (c=1.2, CHCl₃).

Compound 11b. To a stirred solution of amidoalcohol 5b $(0.30 \text{ g}, 0.75 \text{ mmol})$ in DMF (2.5 ml) at 0°C was added imidazole $(0.16 \text{ g}, 2.4 \text{ mmol})$ and TBDMSCl $(0.16 \text{ g},$ 1.1 mmol). The procedure outlined for 11a afforded 0.34 g, (87%) of 11b. ¹H NMR (CDCl₃) δ 7.71 (d, 2H, Ts), 7.32 (m, 5H, Ph), 7.22 (d, 2H, Ts), 5.45 (m, 2H, $CH=CH$), 4.49 (d, $J=2.0$, 2H, PhCH₂), 4.34 (d, $J=8.0$ Hz, 1H, NH), 4.19 (m, 1H), 3.79 (m, 1H), 3.23 (dd, $J=9.0, 6.5, 1H$, 3.19 (dd, $J=9.0, 5.0, 1H$), 2.39 (s, 3H, Ts), $1.20-1.50$ (several m, 4H), 0.92 (m, 12H, t-BuSi+Me), 0.02 (s, 3H, MeSi), -0.03 (s, 3H, MeSi). ¹³C NMR (CDCl₃) δ 143.7, 138.2 (2C), 132.0, 130.3, 129.5 (2C), 128.4, 128.3, 127.7 (2C), 127.5 (2C), 127.1, 74.9 (2C), 71.5, 55.0, 38.3, 36.4, 31.4, 25.8, 25.7, 18.5, 18.2, 13.6, 24.6. IR 3269, 2956, 2928, 2856, 1669, 1455, 1387, 1160, 1094, 814, 697.

Compound 12a. Compound 11a (0.41 g, 0.79 mmol) was dissolved in 21 ml of EtOH. PtO₂ (13 mg, 0.056 mmol) was added and a hydrogen pressure of 1 atm. was applied. The reaction mixture was stirred at RT for 3 h, then filtered through celite and concentrated to give 0.40 mg (97%) of **12a.** ¹H NMR (CDCl₃) δ 7.74 (d, J=8.0 Hz, 2H, Ts), 7.32 $(m, 5H, Ph), 7.25$ (d, $J=8.0$ Hz, 2H, Ts), 4.44 (s, 2H, PhCH2), 3.69 (m, 1H, CHOTBDMS), 3.26 (m, 3H, $CHNHTs+BnOCH₂$), 2.40 (s, 3H, Ts) 1.20-1.60 (several m, 8H), 0.84 (s, 9H, t-BuSi), 0.78 (t, 3H, Me), 0.00 (s, 3H, MeSi), -0.02 (s, 3H, MeSi). ¹³C NMR (CDCl₃) δ 142.8, 138.3, 138.2, 129.4 (2C), 128.2 (2C), 127.5 (2C), 127.4 (2C), 126.9, 74.3, 73.2, 71.2, 54.2, 37.3, 30.54, 30.46, 25.9, 21.5, 18.4, 18.2, 13.9, 24.3, 24.7. IR 3278, 3030, 2955, 2828, 2856, 1496, 1454, 1252, 1161, 1094, 698, 664. HRMS Calcd for $C_{28}H_{45}O_4$ NSiS+H: 520.2939, found: 520.2928. $[\alpha]_D^{22} = -12^{\circ}$ (c=0.8, CHCl₃).

Compound 12b. Compound 11b (0.16 g, 0.3 mmol) was dissolved in 8 ml EtOH. PtO₂ $(5 \text{ mg}, 0.02 \text{ mmol})$ was added and a hydrogen pressure of 1 atm. was applied. The reaction was stirred at RT for 20 min, then filtered through celite and concentrated to give 0.15 g, (92%) of 12b. ¹H NMR (CDCl₃) δ 7.79 (d, J=8.0 Hz, 2H, Ts), 7.40–7.25 $(m, 7H, Ts$ and Ph), 4.39 (s, 2H, PhCH₂), 4.32 (d, J=8.2 Hz, 1H, NH), 3.73 (m, 1H, CHOTBDMS), 3.33 $(dd, J=9.5, 5.5 Hz, 1H), 3.26 (m, 1H), 3.25 (dd, J=9.5,$ 5.5 Hz, 1H), 2.42 (s, 3H, Ts) $1.20-1.60$ (several m, 8H), 0.89 (s, 9H, t-BuSi), 0.79 (t, 3H, Me), 0.03 (s, 3H, MeSi), 0.02 (s, 3H, MeSi). ¹³C NMR (CDCl₃) δ 143.0, 138.5, 132.0, 130.3 (2C), 129.5 (2C), 128.3 (2C), 127.5 (2C), 127.1, 75.0, 73.3, 71.5, 55.1, 38.3, 36.4, 31.4, 25.8, 21.5, 18.5, 18.2, 13.6, 24.7, 24.9. IR 3276, 3030, 2955, 2928, 2856, 1496, 1454, 1252, 1160, 1094, 819, 698.

Compound 13a. Compound 12a (80 mg, 0.15 mmol) was dissolved in MeOH (3 ml), and 5% palladium on carbon

(33 mg) was added. A hydrogen pressure of 1 atm. was applied, and the reaction was stirred at RT for 1 h. After filtration and concentration 60 mg $(93%)$ of compound 13a was recovered. ¹H NMR (CDCl₃) δ 7.74 (d, J=8.4 Hz, 2H, Ts), 7.28 (d, $J=8.4$ Hz, 2H, Ts), 4.55 (d, $J=8.4$, 1H), 3.60 (m, 1H), 3.39 (m, 2H), 3.20 (m, 2H, BnOCH₂), 2.41 (s, 3H, Ts), 1.90 (br s, $1H$) $1.20-1.60$ (several m, $8H$), 0.86 (m, $9H$, t-BuSi), 0.76 (t, 3H, Me), 0.04 (s, 3H, MeSi), 0.01 (s, 3H, MeSi). ¹³C NMR (CDCl₃) δ(143.0, 138.2, 129.4 (2C), 126.9 (2C), 72.5, 66.0, 54.1, 37.3, 30.7, 29.6, 25.9, 21.6, 18.6, 18.1, 13.9, 24.3, 24.4. IR 3510, 3285, 2956, 2929, 2857, 1599, 1462, 1322, 1254, 1159, 1094, 836. HRMS Calcd for $C_{21}H_{39}O_4$ NSiS + Na: 452.2267, found: 452.2265. $[\alpha]_D^{22} = -6^\circ$ (c=1.3, CHCl₃). The product was of >99.5% ee as determined by HPLC analysis using a Chiragel OD-H column (flow rate 0.5 ml/min, 90/10, hexane/2-propanol).

Compound 13b. Compound 12b (147 mg, 0.28 mmol) was dissolved in MeOH (5 ml), and 5% palladium on carbon (101 mg) was added. A hydrogen pressure of 1 atm. was applied, and the reaction was stirred at RT for 3.5 h. After filtration and concentration 110 mg (90%) of compound 13b was recovered. ¹H NMR (CDCl₃) δ 7.75 (d, 2H, Ts), 7.29 (d, 2H, Ts), 4.21 (d, $J=8.0$, 1H), 3.45 (m, 1H), 3.36 (m, 1H), 3.24 (m, 1H), 3.14 (m, 1H), 2.42 (s, 3H, Ts), 1.69 (m, 1H) 1.20±1.50 (several m, 7H), 0.9 (m, 9H, t-BuSi), 0.79 (t, $J=7.0$, 3H, Me), 0.05 (s, 3H, MeSi), 0.02 (s, 3H, MeSi). 13° C NMR (CDCl₃) δ 143.2, 138.6, 129.6 (2C), 127.1 (2C), 72.5, 66.2, 53.9, 37.4, 30.6, 29.4, 25.8, 21.4, 18.5, 18.1, 13.8, 24.2, 24.9. IR 3510, 3282, 2956, 2929, 2857, 1599, 1460, 1322, 1254, 1159, 1094, 835.

Compound 15a. To a stirred solution of alcohol 13a $(0.162 \text{ g}, 0.38 \text{ mmol})$ in THF (4.5 ml) at 0°C , were added Et₃N (109 μ 1, 0.76 mmol) and MsCl (47 μ 1, 0.60 mmol). The resulting solution was stirred at 0° C for 1 h and at RT for 2 h. The reaction mixture was partitioned between ether and water, and the organic phase was dried (Na_2SO_4) . Concentration afforded the crude mesylate 14a, which was dissolved in MeOH (7 ml). K_2CO_3 (0.46 g, 3.3 mmol) was added and the mixture was stirred at RT for 23 h. The solvent was evaporated and $H₂O$ (10 ml) was added, this water phase was extracted EtOAc $(3\times15 \text{ ml})$. Drying and concentration afforded an oil, which was purified by flash chromatography $(SiO₂, pentane:ether (75:25))$ to afford 0.13 g, (81%) of the product 15a. ¹H NMR (CDCl₃) δ 7.79 (d, J=8.0 Hz, 2H, Ts), 7.21 (d, J=8.0 Hz, 2H, Ts), 3.85 (m, 1H, CHN), 3.72-3.69 (m, 3H, CH₂N+ CHOTBDMS), 3.14 (dd, 1H, $J=1.8$, 13.9), 2.38 (s, 3H, Ts), 1.95 (m, 1H), 1.10-1.70 (several m, 7H), 0.85 (m, 12H, t -BuSi+Me), 0.01 (s, 3H, MeSi), 0.00 (s, 3H, MeSi). ¹³C NMR (CDCl₃) δ 142.2, 138.4, 129.0 (2C), 127.4 (2C), 64.6, 52.5, 46.6, 31.1, 26.1, 25.9, 21.5, 20.9, 19.9, 18.4, 14.1, 24.8 (2C). IR 3428, 2956, 2857, 1463, 1327, 1148, 1094, 835. HRMS Calcd for $C_{21}H_{37}O_3$ NSiS+H: 412.2342, found: 412.2232. $[\alpha]_D^{22} = -8^\circ$ ($c=1.1$, CHCl₃).

Compound 15b. The procedure outlined for 15a was employed. To $13b$ (0.141 g, 0.33 mmol) in THF (4 ml) at 0 \degree C, was added Et₃N (90 μ l, 0.65 mmol) and MsCl (40 μ l, 0.52 mmol) and the resulting crude product was dissolved in MeOH (6 ml) and K_2CO_3 (400 mg, 2.9 mmol) was added, this afforded 0.107 g (79%) of product 15b. ¹H NMR

 $(CDCl_3)$ δ 7.72 (d, 2H, Ts), 7.30 (d, 2H, Ts), 3.97 (m, 1H), 3.69 (dd, $J=13.5$, 5.0 Hz, 1H), 3.33 (m, 1H), 2.71 (dd, $J=13.5$, 10.7 Hz, 1H), 2.45 (m, 1H), 2.42 (s, 3H, Ts), $1.20-1.70$ (several m, 8H), 0.91 (m, 12H, t -BuSi+Me), 0.04 (s, 3H, MeSi), 0.01 (s, 3H, MeSi). ¹³C NMR (CDCl₃) ^d 142.9, 138.9, 129.5 (2C), 127.6 (2C), 66.8, 51.8, 46.6, 31.6, 29.0, 26.7, 25.7, 21.4, 19.6, 18.0, 13.8, 24.7, 24.9 (2C). IR 3286, 2956, 2857, 1355, 1175, 1160, 959, 914, 837, 665.

Compound 16a. A solution of 15a (93 mg, 0.23 mmol) in EtOH (2 ml) was added to liquid ammonia (6 ml) with stirring at -78° C. To this mixture was added sodium (0.3 g, 13 mmol) in small portions. The reaction was allowed to reach RT. After 24 h the mixture was diluted with ether (15 ml) and quenched with aqueous NH₄Cl (sat) (1 ml) . The aqueous phase was separated, backextracted with ether (2×15 ml) and dried over Na₂SO₄. Concentration in vacuo afforded 50 mg (85%) of 16a as the only product. This colorless oil was used directly in the next step. ¹H NMR (CDCl₃) δ 3.55 (m, 1H), 3.10 (m, 1H), 2.5 (m, 2H) 1.90 (m, 1H), 1.70 (m, 2H) 1.20 -1.00 (several m, 6H), 0.90 (m, 12H, t -BuSi+Me), 0.00 (s, 6H, MeSi). ¹³C NMR $(CDCl_3)$ δ 69.4, 55.6, 54.5, 38.9, 34.7, 31.6, 25.9, 19.5, 18.2, 14.3, -4.5 . GCMS (CI, CH₄) 258 (M+1, 100%), 242 (M-CH₃, 89%), 214 (M-C₃H₇, 16%), 200 (M-t-Bu, 31%), 126 (M2OTBDMS, 55%). IR 3339, 2956, 2928, 2857, 1463, 1257, 1098. HRMS Calcd for $C_{14}H_{31}ONSi + H$: 258.2254, found: 258.2255. $[\alpha]_D^{22} = +3^\circ$ (c=0.9, CHCl₃).

Compound 16b. To a solution of 15b (100 mg, 0.24 mmol) and $Na₂HPO₄$ (287 mg, 2.0 mmol) in dry MeOH (2 ml) at RT was added Na(Hg) (6%, 796 mg). The mixture was heated to 50° C and stirred for 18 h. The reaction was allowed to reach RT, filtered through celite and the filter cake rinsed with EtOAc. Brine was added and the organic layer was separated. The aqueous phase was backextracted three times with EtOAc, and the combined organic layers were dried (Na₂SO₄). Concentration afforded 50 mg (81%) of 16b. ¹H NMR (CDCl₃) δ 3.71 (m, 1H), 2.85 (dd, J=13.0, 2.5 Hz, 1H), 2.70 (dd, $J=13.0$, 1.7 Hz, 1H) 2.41 (m, 1H), $1.72-1.15$ (several m, 11H), 0.85 (m, 12H, t-BuSi+Me), 0.03 (s, 3H, MeSi) 0.01 (s, 3H, MeSi). ¹³C NMR (CDCl₃) ^d 65.2, 55.5, 52.7, 39.3, 31.7, 27.2, 25.8, 19.1, 18.0, 14.2, $-4.8, -4.9.$

 $(+)$ -Pseudoconhydrine 3. The silyl ether 16a (25 mg, 0.10 mmol) was stirred for 2 h in 3.5 ml of a 1% solution of HCl in EtOH (95%) (10 equiv. of HCl). After concentration, the hydrochloride salt of 3 was isolated and washed with CH₂Cl₂ to give 17 mg (98%). $[\alpha]_D^{21} = +3.67^\circ$ (c=0.52, MeOH). lit.^{6f} $[\alpha]_D^{21} = +3.56^\circ$ (c=0.40, MeOH). ¹H NMR (CD₃OD) δ 3.82 (m, 1H), 3.33 (m, 1H), 3.07 (m, 1H) 2.73 (t, $J=11$ Hz, 1H), 2.10 (m, 2H), 1.68–1.36 (m, 6H), 0.98 (t, $J=7.3$ Hz 3H). Treatment of the salt with 2 M NaOH (1 ml) and extraction with ether $(3 \times 1.5 \text{ ml})$ released the free amine 3, which after removing of the solvent was obtained as a crystalline compound. $[\alpha]_D^{22} = +11.2^{\circ} (c=0.23, \text{EtOH}).$ (lit.^{6e} $[\alpha]_D^{29}$ = +11.1^o(c=1.0, EtOH). Spectral data were in accordance with those previously reported.^{6e}

 (\pm) -epi-Pseudoconhydrine 9. The silyl ether 16b (5 mg, 0.02 mmol) was stirred for 2 h in 0.7 ml of a 1% solution of HCl in EtOH (95%) (10 equiv. of HCl). After concentration the hydrochloride salt of 9 was isolated in 94% yield (3.3 mg). Spectral data of the hydrochloride salt were in accordance with those reported in Ref. 6b. Treatment with 2 M NaOH (1 ml) and extraction with ether $(3\times1.5 \text{ ml})$ gave the free amine 9.

Acknowledgements

Financial support from the Swedish Natural Science Research Council is gratefully acknowledged. We also thank Ms. Mikaela Recknagel for valuable help in the lab.

References

1. (a) Strunzand, G. M.; Finlay, J. A. The alkaloids; Brossi, A., Ed.; Academic Press: San Diego, 1986; Vol. 26, p 89. (b) Jones, T. H.; Blum, M. S.; Robertsson, H. G. J. Nat. Prod. 1990, 53, 429. 2. (a) Christofidis, I.; Welter, A.; Jadot, J. Tetrahedron 1977, 33, 977. (b) Hasseberg, H.-A.; Gerlach, H. Liebigs. Ann. Chem. 1989, 255.

3. (a) Rice, W. Y.; Coke, J. L. J. Org. Chem. 1966, 31, 1010. (b) Highet, R. J. J. Org. Chem. 1964, 29, 471. (c) Highet, R. J.; Highet, P. F. J. Org. Chem. 1966, 31, 1275. (d) Hill, R. K. Chemistry of the Alkaloids; Pelletier, S. W., New York, 1970, pp 385. (e) Toyooka, N.; Yoshida, Y.; Yotsui, Y.; Momose, T. J. Org. Chem. 1999, 64, 4914.

4. Ladenburg, A.; Adam, G. Chem. Ber. 1891, 24, 1671.

5. (a) Marion, L.; Cockburn, W. F. J. Am. Chem. Soc. 1949, 49, 3402. (b) Harding, K. E.; Burks, S. R. J. Org. Chem. 1984, 49, 40. (c) Späth, E.; Kuffner, F.; Ensfeller, L. Chem. Ber. 1933, 66, 591. (d) Brown, E.; Lavoue, J.; Dhal, R. Tetrahedron 1973, 29, 455. 6. (a) Tadano, K. I.; Iimura, Y.; Suami, T. J. Carbohydrate Chem.

1985, 4, 129. (b) Takahata, H.; Inose, K.; Momose, T. Heterocycles 1994, 38, 269. (c) Sakagami, H.; Kamikubo, T.; Ogasawara, K. J. Chem. Soc., Chem. Commun. 1996, 1433. (d) Cossy, J.; Dumas, C.; Pardo, D. G. Synlett, 1997, 905. (e) Moody, C. J.; Lightfoot, A. P.; Gallagher, P. T. J. Org. Chem. 1997, 62, 746. (f) Oppolzer, W.; Bochet, C. G. Tetrahedron Lett. 1995, 36, 2959. (g) Hirai, Y.; Shibuya, K.; Fukuda, Y.; Yokoyama, H.; Yamaguchi, S. Chem. Lett. 1997, 3, 221. (h) Dockner, M.; Sasaki, N. A.; Riche, C.; Potier, P. Liebigs Ann., Recl. 1998, 6, 1267. (i) Herdeis, C.; Schiffer, T. Synthesis 1997, 1405. (j) Agami, C.; Couty, F.; Lam, H.; Mathieu, H. Tetrahedron 1998, 54, 8783.

7. (a) Pettersson-Fasth, H.; Riesinger, S. W.; Bäckvall, J.-E. J. Org. Chem. 1995, 60, 6091. (b) Riesinger, S. W.; Löfstedt, J.; Pettersson-Fasth, H.; Bäckvall, J.-E. Eur. J. Org. Chem. 1999, 3277. 8. (a) Byström, S. E.; Aslanian, R.; Bäckvall J. E. Tetrahedron Lett. 1985, 26, 1749. (b) Kohrt, J. T.; Gu, J.-X.; Johnson, C. R. J. Org. Chem. 1998, 63, 5088.

9. For Pd(0)-catalyzed reactions of vinylepoxides with nucleophiles see: (a) Tsuji, J.; Kataoka, H.; Kobayashi, Y. Tetrahedron Lett. 1981, 22, 2575. (b) Trost, B. M.; Molander, G. A. J. Am. Chem. Soc. 1981, 103, 5969. (c) Trost, B. M.; Sudhakar, A. R. J. Am. Chem. Soc. 1987, 109, 3792. (d) Bernet, B.; Vasella, A. Tetrahedron Lett. 1983, 24, 5491. (e) Tenaglia, A.; Waegell, B. Tetrahedron Lett. 1988, 29, 4851. (f) Trost, B. M.; Romero, A. G. J. Org. Chem. 1986, 51, 2332. (g) Tsuji, J. Pure & Appl. Chem. 1982, 54, 197. (h) Trost, B. M.; Kou, G.-H.; Benneche, T. J. Am. Chem. Soc. 1988, 110, 621. (i) Trost, B. M.; Shi, Y. J. Am. Chem. Soc. 1993, 115, 9421. (j) Trost, B. M.; Bunt, R. C. Angew. Chem.

1996, 108, 70. Trost, B. M.; Krueger, A. C.; Bunt, R. C.; Zambrano, J. J. Am. Chem. Soc. 1996, 118, 6520.

- 10. Schink, H. E.; Pettersson, H.; Bäckvall, J. E. J. Org. Chem. 1991, 56, 2769.
- 11. Bäckvall, J. E.; Andersson, P. G.; Stone, G. B.; Gogoll, A. J. Org. Chem. 1991, 56, 2988.
- 12. Bäckvall, J. E.; Schink, H. E.; Renko, Z. D. J. Org. Chem. 1990, 55, 826.
- 13. Yamazaki, N.; Kibayashi, C. J. Am. Chem. Soc. 1989, 111, 1396.
- 14. Cunico, R. F.; Bedell, L. J. Org. Chem. 1980, 45, 4797.
- 15. Trost, B. M.; Weber, L.; Strege, P.; Fullerton, T. J.; Dietsche, T. J. J. Am. Chem. Soc. 1978, 100, 3426.
- 16. Schloss, J. E.; Hartman, F. C. Bioorganic Chem. 1980, 9, 217.
- 17. (a) McDonald et al.^{17b} recently published a high yielding

method for the LiAlH₄ reduction of 17 to $18a$, which in our hands gave a lower yield than the method of Schloss and Hartman.16 (b) McDonald, W. S.; Verbicky, C. A.; Zercher, C. K. J. Org. Chem. 1997, 62, 1215.

- 18. Bouzide, A.; Sauvé, G. Tetrahedron Lett. 1997, 34, 5945.
- 19. In the procedure described for monobenzylation of diols 1.5 equiv. of $Ag₂O$ was employed. With our substrate 18a this led to substantial dibenzylation. However, the use of only 1 equiv. of Ag_2O significantly improved the yield of 18b.
- 20. (a) Katsuki, T.; Lee, A. W. M.; Ma, P.; Martin, V. S.; Masamune, S.; Sharpless, K. B.; Tuddenham, D.; Walker, F. J. J. Org. Chem. 1982, 47, 1373. (b) Gao, Y.; Hanson, R. M.; Klunder, J. M.; Ko, S. Y.; Masamune, S.; Sharpless, K. B. J. Am. Chem. Soc. 1987, 109, 5765.